

Reaction of *N*-(4-Pyridylmethyl)benzamide *N*-Oxides and  
*N*-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]benzamides with  
 1,3-Diphenyl-1,3-propanedione

Miguel F. Braña, María L. López Rodríguez\* and José M. Castellano

Departamento de Química Orgánica I. Facultad de Ciencias Químicas, Universidad Complutense,  
 28040-Madrid, Spain

Purificación Fernández and Amando Garrido-Pertierra

Departamento de Bioquímica, Facultad de Veterinaria, Universidad Complutense,  
 28040-Madrid, Spain

Received July 11, 1988

Reaction of *N*-(4-pyridylmethyl)benzamide *N*-oxides **2** with 1,3-diphenyl-1,3-propanedione in the presence of acetic anhydride afforded 1,1-dibenzoyl-2-(4-pyridyl)-2-(benzoylamino)ethanes **4** in low yield. Treatment of *N*-[( $\alpha$ -acetoxy)-4-pyridylmethyl]benzamides **3** with 1,3-diphenyl-1,3-propanedione in the presence of triethylamine and chloroform as a solvent provided **4** in high yield. Reaction of **4** with nucleophiles as hydrazine, methyl and phenylhydrazine gave the corresponding pyrazoles **5**.

*J. Heterocyclic Chem.*, **27**, 401 (1990).

Picobenzide **1a** [1] is a potent pharmacological agent. During the last few years, we have obtained new derivatives [2-13] of **1a** by various modifications of its skeleton. As a part of our program, we have shown that there are two methods to functionalize the methylene group of **1a**. The first consists of the reaction of the *N*-oxide **2a** with active methylene compounds in the presence of the acetic anhydride. The second method utilises the *N*-[( $\alpha$ -acetoxy)-4-pyridylmethyl] derivative **3a** with acidic compounds.

These facts, and our continuing interest in the chemistry of Picobenzide, prompted us to investigate the reactivity of a series of *N*-(4-pyridylmethyl)benzamide *N*-oxides **2** and *N*-[( $\alpha$ -acetoxy)-4-pyridylmethyl]benzamides **3** with 1,2-diphenyl-1,3-propanedione.

The starting *N*-(4-pyridylmethyl)benzamide *N*-oxides **2**

were prepared by oxidation of the corresponding benzamides with hydrogen peroxide in acetic acid [5]. Benzamides **1** were obtained by the direct reaction of the appropriate acids with 4-aminomethylpyridine in the presence of dicyclohexylcarbodiimide [14].

Reaction of *N*-(4-pyridylmethyl)-3,5-dimethylbenzamide *N*-oxide **2a** with 1,3-diphenyl-1,3-propanedione in the presence of acetic anhydride afforded 1,1-dibenzoyl-2-(4-pyridyl)-2-(3,5-dimethylbenzoylamino)ethane **4a** in 20% yield. This structure was established on the basis of elemental analysis and spectral data (Tables I-III).

Similarly, the *N*-oxides **2b-f** react with 1,3-diphenyl-1,3-propanedione to give the corresponding ethane derivatives **4b-f** in low yield (< 20% in all cases, method A). The synthetic results and various spectral data are summarized in Tables I-III.

Table I

1,1-Dibenzoyl-2-(4-pyridyl)-2-(benzoylamino)ethanes **4a-f**.

Compound	R	Yield %		Mp [a]	Formula	Analyses %		
		Method A	Method B			Calcd./Found	C	H
<b>4a</b>	3,5(CH <sub>3</sub> ) <sub>2</sub>	20	84	204-205°	C <sub>30</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	77.89 78.11	5.66 5.41	6.05 6.21
<b>4b</b>	H	19	88	211-212°	C <sub>28</sub> H <sub>22</sub> N <sub>2</sub> O <sub>3</sub>	77.39 77.40	5.10 4.80	6.45 6.64
<b>4c</b>	2-CH <sub>3</sub>	15	49	200-202° [b]	C <sub>27</sub> H <sub>24</sub> N <sub>2</sub> O <sub>3</sub>	77.65 77.51	5.39 5.47	6.24 6.34
<b>4d</b>	4-OCH <sub>3</sub>	17	72	216-217°	C <sub>27</sub> H <sub>24</sub> N <sub>2</sub> O <sub>4</sub>	74.97 75.10	5.20 5.29	6.03 6.32
<b>4e</b>	4-NO <sub>2</sub>	12	40	215-216°	C <sub>26</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub>	70.13 70.24	4.41 4.45	8.76 8.61
<b>4f</b>	4-Cl	17	77	214-215°	C <sub>26</sub> H <sub>21</sub> ClN <sub>2</sub> O <sub>3</sub>	77.71 71.79	4.51 4.39	5.97 6.03

[a] From ethanol. [b] From benzene.

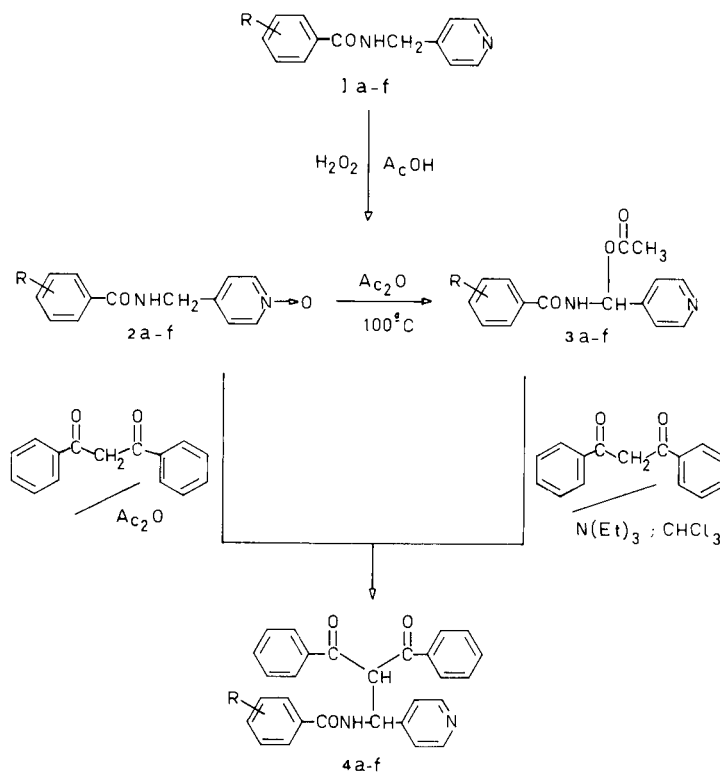
Table II

IR [a] and <sup>1</sup>H NMR [b] Spectral Data of Compounds 4a-f

Compound	R	$\nu$ NH	IR, cm <sup>-1</sup> $\nu$ CO	<sup>1</sup> H NMR, $\delta$
4a	3,5-(CH <sub>3</sub> ) <sub>2</sub>	3300	1680, 1660, 1640	2.2 (s, 6H, 2CH <sub>3</sub> ), 5.9 (t, 1H, CH), 6.6 (d, 1H, CHO), 6.8-7.6 (m, 11H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, H <sub>2</sub> , H <sub>4</sub> , H <sub>6</sub> -PhCONH, H <sub>3</sub> , H <sub>5</sub> -Py), 7.6-8.1 (m, 4H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py), 8.6 (d, 1H, NH)
4b	H	3360	1690, 1670, 1640	6.1 (t, 1H, CH), 6.7 (d, 1H, CHCO), 7.1-7.6 (m, 13H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, 5H-PhCONH, H <sub>3</sub> , H <sub>5</sub> -Py), 7.7-8.2 (m, 4H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO), 8.4 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py), 8.9 (d, 1H, NH)
4c	2-CH <sub>3</sub>	3300	1680, 1660, 1630	2.0 (s, 3H, CH <sub>3</sub> ), 5.9 (t, 1H, CH), 6.4 (d, 1H, CHCO), 6.8-7.4 (m, 12H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, 4H-PhCONH, H <sub>3</sub> , H <sub>5</sub> -Py), 7.5-7.9 (m, 4H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py), 8.6 (d, 1H, NH)
4d	4-OCH <sub>3</sub>	3320	1690, 1660, 1630	3.6 (s, 3H, OCH <sub>3</sub> ), 5.9 (t, 1H, CH), 6.5 (d, 1H, CHCO), 6.6-7.5 (m, 12H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, 4H-PhCONH, H <sub>3</sub> , H <sub>5</sub> -Py), 7.6-8.0 (m, 4H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py), 8.5 (d, 1H, NH)
4e	4-NO <sub>2</sub>	3340	1970 (overlapped)	5.9 (t, 1H, CH), 6.5 (d, 1H, CHCO), 7.2-7.5 (m, 8H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, H <sub>3</sub> , H <sub>5</sub> -Py), 7.6-8.4 (m, 10H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO, 4H-PhCONH, H <sub>2</sub> , H <sub>6</sub> -Py), 9.0 (d, 1H, NH)
4f	4-Cl	3320	1680, 1660, 1630	5.9 (t, 1H, CH), 6.5 (d, 1H, CHCO), 7.0-7.5 (m, 12H, 2H <sub>3</sub> , 2H <sub>4</sub> , 2H <sub>5</sub> -PhCO, 4H-PhCONH, H <sub>3</sub> , H <sub>5</sub> -Py), 7.6-8.0 (m, 4H, 2H <sub>2</sub> , 2H <sub>6</sub> -PhCO), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py), 8.8 (d, 1H, NH)

[a] Potassium Bromide. [b] DMSO-d<sub>6</sub>.

Scheme 1

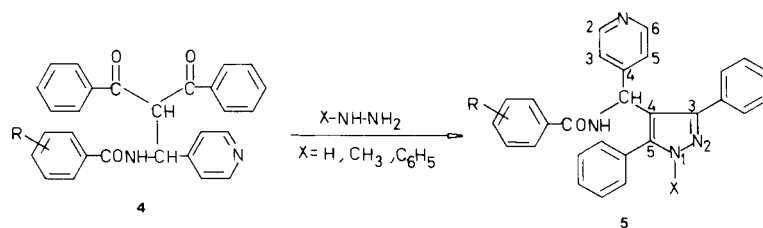
a, R = 3,5 (CH<sub>3</sub>)<sub>2</sub>

b, R = H

c, R = 2-CH<sub>3</sub>d, R = 4-OCH<sub>3</sub>e, R = 4-NO<sub>2</sub>

f, R = 4-Cl

Scheme 2



a, R = 3,5-(CH <sub>3</sub> ) <sub>2</sub>	a, R = 3,5-(CH <sub>3</sub> ) <sub>2</sub>	X = H
c, R = 2-CH <sub>3</sub>	a', R = 3,5-(CH <sub>3</sub> ) <sub>2</sub>	X = C <sub>6</sub> H <sub>5</sub>
f, R = 4-Cl	c, R = 2-CH <sub>3</sub>	X = CH <sub>3</sub>
	c', R = 2-CH <sub>3</sub>	X = C <sub>6</sub> H <sub>5</sub>
	f, R = 4-Cl	X = C <sub>6</sub> H <sub>5</sub>

Table III

<sup>13</sup>C NMR of **4a-f** (DMSO-d<sub>6</sub>, δ)

Compound	R	CH <sub>3</sub>	CH	C <sub>3</sub> and C <sub>5</sub> pyridine	Ph	C <sub>4</sub> pyridine	C <sub>2</sub> and C <sub>6</sub> pyridine	CONH	C=O
<b>4a</b>	3,5-(CH <sub>3</sub> ) <sub>2</sub>	20.7	52.9, 58.4	122.9	124.9, 128.5, 128.8, 129.0, 132.7, 134.1, 135.5, 136.2, 137.4	149.1	149.6	166.3	193.6
<b>4b</b>	H	—	52.9, 58.5	122.8	127.1, 128.3, 128.5, 128.6, 129.0, 131.5, 133.9, 134.1, 135.4, 136.1	148.9	149.5	165.9	193.4 193.7
<b>4c</b>	2-CH <sub>3</sub>	19.0	52.6, 58.7	122.8	125.5, 128.5, 128.7, 129.0, 129.5, 130.5, 134.1, 135.5, 136.1, 136.5	149.0	149.6	168.4	193.3 193.7
<b>4d</b>	4-OCH <sub>3</sub>	55.4	52.9, 58.6	122.9	113.6, 126.1, 128.7, 129.1, 134.1, 134.2, 135.5, 136.1, 161.9	149.2	149.6	165.4	193.5 193.7
<b>4e</b>	4-NO <sub>2</sub>	—	53.1, 58.5	123.5	122.9, 128.5, 128.7, 129.0, 134.1, 135.4, 136.0, 139.5, 139.6	148.5	149.2	164.4	193.4 193.7
<b>4f</b>	4-Cl	—	53.0, 58.5	122.9	128.4, 128.7, 129.0, 129.1, 132.6, 134.1, 135.5, 136.1, 136.5	148.8	149.6	165.0	193.4 193.7

Table IV

4-((Benzoylamino)(4-pyridyl))methyl-3,5-diphenylpyrazoles **5**

Compound	R	X	Yield %	Mp [a]	Formula	Analyses, %		
						Calcd./	Found	N
<b>5a</b>	3,5-(CH <sub>3</sub> ) <sub>2</sub>	H	60	220-221° [b]	C <sub>30</sub> H <sub>26</sub> N <sub>4</sub> O	78.57	5.71	12.21
						78.82	5.74	11.98
<b>5a'</b>	3,5-(CH <sub>3</sub> ) <sub>2</sub>	C <sub>6</sub> H <sub>5</sub>	87	162-163°	C <sub>30</sub> H <sub>30</sub> N <sub>4</sub> O	80.87	5.65	10.48
						80.61	5.63	10.36
<b>5c</b>	2-CH <sub>3</sub>	CH <sub>3</sub>	83	138-139°	C <sub>30</sub> H <sub>26</sub> N <sub>4</sub> O	78.57	5.71	12.22
						78.31	5.77	12.13
<b>5c'</b>	2-CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	77	204-205°	C <sub>35</sub> H <sub>28</sub> N <sub>4</sub> O	80.74	5.42	10.76
						80.59	5.40	10.77
<b>5f</b>	4-Cl	C <sub>6</sub> H <sub>5</sub>	78	183-184°	C <sub>34</sub> H <sub>25</sub> ClN <sub>4</sub> O	75.47	4.65	10.35
						75.36	4.63	10.23

[a] From ethanol-water. [b] From benzene.

Table V  
 IR [a] and <sup>1</sup>H NMR [b] Spectral Data for Compounds 5

Compound	R	X	IR, cm <sup>-1</sup>		<sup>1</sup> H NMR, δ
			ν NH	ν CO	
5a	3,5-(CH <sub>3</sub> ) <sub>2</sub>	H	2800-3300	1640	2.2 (s, 6H, 2CH <sub>3</sub> ), 6.5-7.4 (m, 18H, CH, 13H-Ph, H <sub>3</sub> , H <sub>5</sub> -Py, 2NH), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py)
5a'	3,5-(CH <sub>3</sub> ) <sub>2</sub>	C <sub>6</sub> H <sub>5</sub>	3300	1660	2.3 (s, 6H, 2CH <sub>3</sub> ), 6.7-7.6 (m, 22H, CH, 18H-Ph, H <sub>3</sub> , H <sub>5</sub> -Py, NH), 8.5 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py)
5c	2-CH <sub>3</sub>	CH <sub>3</sub>	3300	1660	2.2 (s, 3H, CH <sub>3</sub> ), 3.7 (s, 3H, CH <sub>3</sub> N), 6.3-7.5 (m, 18H, CH, 14H-Ph, H <sub>3</sub> , H <sub>5</sub> -Py, NH), 8.1 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py)
5c'	2-CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	3300	1660	2.2 (s, 3H, CH <sub>3</sub> ), 6.2-7.4 (m, 23H, CH, 19H-Ph, H <sub>3</sub> , H <sub>5</sub> -Py, NH), 8.2 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py)
5f	4-Cl	C <sub>6</sub> H <sub>5</sub>	3200	1660	6.4-7.4 (m, 23H, CH, 19H-Ph, H <sub>3</sub> , H <sub>5</sub> -Py, NH), 8.3 (d, 2H, H <sub>2</sub> , H <sub>6</sub> -Py)

[a] Potassium Bromide. [b] Deuteriochloroform.

 Table VI  
<sup>13</sup>C NMR of 5 (DMSO-d<sub>6</sub>, δ)

Compound	R	X	CH <sub>3</sub>	CH	C <sub>4</sub>		Ph	C <sub>4</sub> pyridine	C <sub>2</sub> and C <sub>6</sub> pyridine	C <sub>3</sub> and C <sub>5</sub> pyrazole	CONH
					pyrazole	pyridine					
5a [b]	3,5-(CH <sub>3</sub> ) <sub>2</sub>	H	20.9	48.8	113.9	122.2	122.5, 128.4, 128.6, 132.9, 133.9, 137.5	148.9	150.0	151.1	166.6
5a'	3,5-(CH <sub>3</sub> ) <sub>2</sub>	C <sub>6</sub> H <sub>5</sub>	20.8	47.2	117.4	121.4	124.1, 124.5, 127.0, 128.3, 128.5, 128.6, 128.8, 128.9, 129.3, 129.7, 132.9, 133.1, 137.9, 139.1, 141.7	150.7	149.7	151.4	166.1
5c	2-CH <sub>3</sub>	CH <sub>3</sub>	19.4 36.6	47.5	115.3	121.0	125.0, 126.0, 127.8, 128.2, 128.4, 128.8, 129.2, 129.6, 130.6, 132.7, 134.7, 135.9, 142.6	148.8	148.8	150.5	168.3
5c'	2-CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	19.7	47.4	117.3	121.3	124.5, 125.3, 126.2, 127.0, 128.3, 128.5, 128.6, 128.9, 129.2, 129.3, 129.7, 130.0, 131.0, 132.8, 134.6, 136.5, 139.2, 142.1	150.8	149.6	150.9	168.4
5f	4-Cl	C <sub>6</sub> H <sub>5</sub>	—	47.5	117.0	121.4	124.9, 127.0, 127.8, 128.3, 128.6, 128.7, 128.9, 129.3, 129.6, 131.4, 132.9, 137.6, 139.0, 141.7	150.6	149.6	150.8	164.7

[b] Deuteriochloroform.

We have previously shown [7-11] that the *N*-[(α-acetoxy)-4-pyridylmethyl]-3,5-dimethylbenzamide **3a** is the intermediate in the reaction of the *N*-oxide **2a** with hydrogen acidic compounds. So, in order to get **4a-f** in better yield, we have carried out the reaction of **3a-f** with 1,3-diphenyl-1,3-propanedione in the presence of triethylamine using chloroform as a solvent. This procedure afforded **4a-f** in high yield (method B) (Tables I-III).

The starting *N*-[(α-acetoxy)-4-pyridylmethyl]benzamides **3a-f** were obtained by the direct reaction of the corresponding *N*-oxides **2a-f** with acetic anhydride at 100° [4].

Finally, owing to our interest in the synthesis of hetero-

cyclic systems related to Picobenzide, we have carried out the reaction of **4** with nucleophiles as hydrazine, methyl and phenylhydrazine. These reactions lead to the formation of the corresponding pyrazoles **5**. The synthetic results and various spectral data are summarized in Tables IV-VI.

#### EXPERIMENTAL

Melting points were determined on a Büchi 510D apparatus in open capillaries and are uncorrected. The ir spectra were measured on a Perkin-Elmer 781 spectrophotometer. The <sup>1</sup>H nmr spectra were recorded on a Varian T-60A spectrometer using TMS as an internal standard. The

<sup>13</sup>C nmr spectra were obtained on a Varian FT-80A spectrometer. Mass spectrometry was performed with a Varian MAT-711 apparatus. Elemental analyses were performed by the Consejo Superior de Investigaciones Científicas, Madrid.

N-(4-Pyridylmethyl)benzamides **1a-f** [14] and N-(4-pyridylmethyl)benzamide N-oxides **2a-f** [5] were obtained according to literature methods.

General Procedure for N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]benzamides **3a-f**.

A solution of the corresponding N-(4-pyridylmethyl)benzamide N-oxide **2a-f** (0.02 mole) in 50 ml of acetic anhydride was heated at 100° for 20 minutes. The solution was cooled to room temperature and compounds **3** were isolated by crystallization of the precipitate formed or by evaporation under reduced pressure and trituration of the residual oil with ether.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]-3,5-dimethylbenzamide **3a** [4] was prepared according to the previously reported procedure.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]benzamide (**3b**).

This compound was obtained in a yield of 40%, mp 104-105° (benzene); ir (potassium bromide): 3260 (NH), 1730 (C=O, ester), 1640 (C=O, amide); <sup>1</sup>H nmr (DMSO-d<sub>6</sub>):  $\delta$  2.2 (s, 3H, COCH<sub>3</sub>), 6.7 (d, 1H, CH), 7.0-8.0 (m, 7H, 5H-phenyl, H<sub>3</sub> and H<sub>5</sub>-pyridine), 8.5 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-pyridine, J = 5 Hz), 9.6 (d, 1H, NH).

Anal. Calcd. for C<sub>15</sub>H<sub>14</sub>N<sub>2</sub>O<sub>3</sub>: C, 66.65; H, 5.22; N, 10.36. Found: C, 66.48; H, 5.16; N, 10.46.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]-2-methylbenzamide (**3c**).

This compound was obtained in a yield of 80%, mp 101-102° (ether-petroleum ether); ir (potassium bromide): 3300 (NH), 1730 (C=O, ester), 1640 (C=O, amide); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.1 (s, 3H, COCH<sub>3</sub>), 2.3 (s, 3H, CH<sub>3</sub>), 6.9-7.4 (m, 7H, CH, 4H-phenyl, H<sub>3</sub> and H<sub>5</sub>-pyridine), 7.9 (d, 1H, NH), 8.2 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-pyridine, J = 5 Hz).

Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>: C, 67.59; H, 5.67; N, 9.85. Found: C, 67.63; H, 5.56; N, 9.80.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]-4-methoxybenzamide (**3d**).

This compound was obtained in a yield of 88%, mp 108-109° (benzene); ir (potassium bromide): 3260 (NH), 1730 (C=O, ester), 1650 (C=O, amide); <sup>1</sup>H nmr (DMSO-d<sub>6</sub>):  $\delta$  2.1 (s, 3H, COCH<sub>3</sub>), 3.7 (s, 3H, OCH<sub>3</sub>), 6.4 (d, 1H, CH), 6.8 (d, 2H, H<sub>3</sub> and H<sub>5</sub>-phenyl J = 9 Hz), 7.3 (d, 2H, H<sub>3</sub> and H<sub>5</sub>-pyridine J = 5 Hz), 7.8 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-phenyl J = 9 Hz), 8.4 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-pyridine J = 5 Hz), 9.3 (d, 1H, NH).

Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>4</sub>: C, 63.98; H, 5.37; N, 9.33. Found: C, 63.90; H, 5.53; N, 9.45.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]-4-nitrobenzamide (**3e**).

This compound was obtained in a yield of 85%, mp 107-108° (ether); ir (potassium bromide): 3280 (NH), 1740 (C=O, ester), 1660 (C=O, amide); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.1 (s, 3H, COCH<sub>3</sub>), 7.0-7.4 (m, 3H, CH, H<sub>3</sub> and H<sub>5</sub>-pyridine), 7.6-8.1 (m, 5H, 4H-phenyl, NH), 8.2 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-pyridine, J = 5 Hz).

Anal. Calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>5</sub>: C, 57.14; H, 4.15; N, 13.32. Found: C, 57.22; H, 4.38; N, 13.48.

N-[( $\alpha$ -Acetoxy)-4-pyridylmethyl]-4-chlorobenzamide (**3f**).

This compound was obtained in a yield of 82%, mp 124-125° (chloroform-petroleum ether); ir (potassium bromide): 3260 (NH), 1730 (C=O, ester), 1660 (C=O, amide); <sup>1</sup>H nmr (DMSO-d<sub>6</sub>):  $\delta$  2.2 (s, 3H, COCH<sub>3</sub>), 7.2-7.6 (m, 5H, CH, H<sub>3</sub> and H<sub>5</sub>-phenyl, H<sub>3</sub> and H<sub>5</sub>-pyridine), 7.9 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-phenyl J = 9 Hz), 8.5 (d, 2H, H<sub>2</sub> and H<sub>6</sub>-pyridine, J = 5 Hz), 9.7 (d, 1H, NH).

Anal. Calcd. for C<sub>15</sub>H<sub>13</sub>ClN<sub>2</sub>O<sub>3</sub>: C, 59.12; H, 4.29; N, 9.19; Cl, 11.63. Found: C, 59.10; H, 4.08; N, 9.15; Cl, 11.49.

General Procedures for 1,1-Dibenzoyl-2-(4-pyridyl)-2-(benzoylamino)ethanes **4a-f**.

Method A.

A solution of the corresponding N-oxide **2a-f** (0.02 mole) and 1,3-diphenyl-1,3-propanedione (0.02 mole) in 50 ml of acetic anhydride was heated at 100° until the completion of the reaction (determined by tlc). The solvent was evaporated under reduced pressure to yield the compounds.

Method B.

To a solution of the corresponding N-[( $\alpha$ -acetoxy)-4-pyridylmethyl]benzamide **3a-f** (0.005 mole) in 25 ml of chloroform was added 1,3-diphenyl-1,3-propanedione (0.005 mole) and 5 ml of triethylamine. The mixture was refluxed until the completion of the reaction (determined by tlc) and the solvent was evaporated under reduced pressure to yield the compounds.

Reaction of 1,1-Dibenzoyl-2-(4-pyridyl)-2-(3,5-dimethylbenzoylamino)ethane (**4a**) and Hydrazine.

A suspension of **4a** (0.924 g, 0.0002 mole) and 80% hydrazine (0.08 g, 0.002 mole) in 30 ml of ethanol was refluxed until the completion of the reaction (determined by tlc). After being cooled, the precipitate was filtered off and crystallized to give 4-[(3,5-dimethylbenzoylamino)(4-pyridyl)methyl]-3,5-diphenylpyrazole **5a**; ms: m/z (relative intensity %) 458 (M<sup>+</sup>, 85), 380 (M<sup>+</sup>-C<sub>2</sub>H<sub>4</sub>N, 22), 325 (M<sup>+</sup>-C<sub>2</sub>H<sub>5</sub>O<sup>+</sup>, 77), 133 (C<sub>6</sub>H<sub>5</sub>O<sup>+</sup>, 105 (C<sub>6</sub>H<sub>5</sub><sup>+</sup>, 44).

Reaction of 1,1-Dibenzoyl-2-(4-pyridyl)-2-(benzoylamino)ethanes **4a,c,f** with Methyl and Phenylhydrazine.

To a suspension of **4** (0.004 mole) in 40 ml of ethanol was added 98% methyl or phenylhydrazine (4 ml) and acetic acid (4 ml). The mixture was refluxed until the completion of the reaction (determined by tlc), and then basified with 10% sodium hydroxide to give **5**.

## REFERENCES AND NOTES

- [1] M. F. Braña, J. M. Castellano, C. M. Roldán and P. F. Rabadán, *Arch. Farmacol. Toxicol.*, **IV**, 273 (1978); M. A. Nalda and S. Carrasco, 7th World Congress of Anaesthesiologist, Hamburg, 1980.
- [2] M. F. Braña, and M. L. López Rodríguez, *Tetrahedron Letters*, **21**, 3923 (1980).
- [3] M. F. Braña, M. L. López Rodríguez and R. López Arenosa, *An. Quim.*, **77C**, 159 (1981).
- [4] M. F. Braña and M. L. López Rodríguez, *J. Heterocyclic Chem.*, **18**, 869 (1981).
- [5] M. F. Braña, M. L. López Rodríguez, J. Garrido and C. M. Roldán, *J. Heterocyclic Chem.*, **18**, 1305 (1981).
- [6] M. F. Braña and M. L. López Rodríguez, *J. Heterocyclic Chem.*, **19**, 1297 (1982).
- [7] M. F. Braña, J. M. Castellano, and M. L. López Rodríguez, *J. Heterocyclic Chem.*, **20**, 1723 (1983).
- [8] M. F. Braña and M. J. R. Yunta, *An. Quim.*, **78C**, 210 (1982); *An. quim.*, **79C**, 439 (1983); *An. Quim.*, **80C**, 160 (1984).
- [9] M. F. Braña, J. M. Castellano and M. J. R. Yunta, *Heterocycles*, **22**, 113 (1984).
- [10] M. F. Braña, J. Dotor, J. M. Castellano and M. L. López Rodríguez, *Real Academia de Ciencias Exactas, Físicas y Naturales de Madrid*, **LXXVIII**, 39, 367 (1984).
- [11] M. F. Braña, M. L. López Rodríguez, C. Rodríguez and A. Garrido-Pertierra, *J. Heterocyclic Chem.*, **23**, 1019 (1986).
- [12] M. F. Braña, J. M. Castellano, M. C. Redondo and J. Esplugues, *J. Heterocyclic Chem.*, **24**, 741 (1987).
- [13] M. F. Braña, J. M. Castellano, M. C. Redondo and M. J. R. Yunta, *J. Heterocyclic Chem.*, **24**, 833 (1987).
- [14] M. F. Braña, M. L. López Rodríguez, J. M. Castellano, R. P. A. Ossorio and C. M. Roldán, UK Patent Appl. GB 2078215A (1982).